

## Structure of 4 $\alpha$ -Carbomethoxy-15 $\alpha$ -cyano-4 $\beta$ -methyl-5 $\alpha$ ,13 $\alpha$ -androstan

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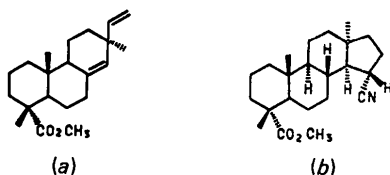
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**Abstract.** C<sub>23</sub>H<sub>35</sub>NO<sub>2</sub>,  $M_r = 357.5$ , orthorhombic,  $P2_12_12_1$ ,  $a = 7.825$  (1),  $b = 14.976$  (2),  $c = 17.702$  (3) Å,  $V = 2074.4$  Å<sup>3</sup>,  $Z = 4$ ,  $D_x = 1.14$  g cm<sup>-3</sup>,  $\lambda(\text{Cu } K\alpha) = 1.54178$  Å,  $\mu = 5.6$  cm<sup>-1</sup>,  $F(000) = 784$ , room temperature, final  $R = 0.038$  for 1182 observed reflections. The bond lengths and angles are normal. The *A/B* and *B/C* junctions are *trans* whereas the *C/D* junction is *cis* with the *D* ring directed towards the  $\beta$  face of the molecule.

**Introduction.** The resin acids extracted from the conifer resins are useful for the synthesis of biologically active compounds. With their tricyclic backbones, they can be considered as potential intermediates for the elaboration of steroid derivatives. A few attempts in this field have been made, starting mainly from abietic or dehydroabietic acids (Steindl & Haslinger, 1985; Curini, Coccia, Ceccherelli, Halls, Porter & Wenkert, 1984; Wenkert *et al.*, 1982; Tahara, Shimayaki, Itoh, Yoshiroto & Masayuki, 1975; Wirthlin, Wehrli, Jeger, 1974; Wheeler & Witt, 1972). In the present work we start with the methyl pimarate (*a*) to provide the basis for the steroid skeleton. During the process the title compound (*b*) was isolated and the present X-ray analysis was undertaken to establish its structure.



**Experimental.** The title compound crystallized at room temperature from a methanol solution. Crystal dimensions 0.1 × 0.1 × 0.3 mm. Enraf–Nonius CAD-4 diffractometer, Cu  $K\alpha$  radiation, graphite mono-

chromator. Lattice parameters from least-squares adjustments to setting angles of 25 reflections with  $21 < 2\theta < 42^\circ$ . Corrections for Lorentz and polarization effects; no corrections for absorption or secondary extinction;  $\omega$ - $2\theta$  scans. Scan width  $(1.0 + 0.14 \tan\theta)^\circ$ .  $\theta_{\text{max}} = 65^\circ$ . Range of *hkl*:  $h: 0 \rightarrow 9$ ;  $k: 0 \rightarrow 17$ ;  $l: 0 \rightarrow 20$ . Intensity variation of three standard reflections  $< 3\%$ . 2033 unique reflections measured; 1121 with  $I > 3\sigma(I)$  considered observed. Solution by direct methods (*MITHRIL*: Gilmore, 1984). Refinement on  $F$  by block-diagonal least squares, anisotropic thermal parameters for C, N and O, isotropic for H atoms (located in  $\Delta F$  map).  $R = 0.038$ ,  $wR = 0.039$ ;  $w = 1/\sigma^2(F_o)$  based on counting statistics;  $S = 1.12$ ;  $\Delta/\sigma(\text{max.}) = 0.02$ . Max. and min. heights in final  $\Delta\rho$  map  $+0.31$  and  $-0.27$  e Å<sup>-3</sup>. Atomic scattering factors from *International Tables for X-ray Crystallography* (1974) for non-H and from Stewart, Davidson & Simpson (1965) for H. Local programs *CRISAFFI* and *CRISUTIL*; Mini-6/92 Bull.

**Discussion.** A projection of the molecular structure with numbering of atoms is shown in Fig. 1.\* Final atomic parameters are listed in Table 1. Bond lengths and angles are listed in Table 2 together with the main torsion angles relevant to the conformation of the molecular backbone.

The intramolecular geometry of the structure is essentially normal. The *A*, *B* and *C* rings all have chair conformations and are slightly flattened in that the torsional angles are less than  $60^\circ$ . This flattening is often observed in steroids belonging to the androstan

\* Lists of structure factors, anisotropic thermal parameters and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 43444 (16 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

family (Weeks, Cooper & Norton, 1971; Precigoux, Busetta, Courseille & Hospital, 1972; Precigoux & Fornies-Marquina, 1973). The *D* ring has a slightly distorted half-chair conformation (Bucourt, 1964). The *A/B* and *B/C* junctions are *trans* whereas the *C/D* junction is *cis* (Fig. 2). In the packing, there is only one intermolecular O...C distance less than the sum of the van der Waals radii: O(44)(*x,y,z*)—C(16)(*x+1, y, z*) = 3.284 (5) Å.

The authors are indebted to M. Saux who made the experimental X-ray measurements.

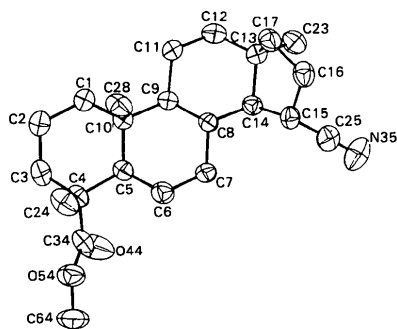


Fig. 1. View of the title compound illustrating atom labelling. Thermal ellipsoids scaled to 50% probability.

Table 1. Final atomic coordinates ( $\times 10^4$ ) and  $B_{eq}$  values with e.s.d.'s in parentheses

	$x$	$y$	$z$	$B_{eq}(\text{Å}^2)$
C(1)	7665 (6)	448 (2)	3161 (2)	4.3 (2)
C(2)	8796 (6)	509 (3)	3866 (2)	5.4 (3)
C(3)	9949 (5)	1322 (3)	3843 (2)	4.8 (2)
C(4)	8938 (5)	2205 (2)	3755 (2)	3.6 (2)
C(5)	7715 (5)	2108 (2)	3059 (2)	3.0 (2)
C(6)	6736 (5)	2971 (2)	2889 (2)	3.9 (2)
C(7)	5917 (5)	2912 (2)	2107 (2)	3.7 (2)
C(8)	4764 (5)	2090 (2)	2025 (2)	2.9 (2)
C(9)	5746 (5)	1231 (2)	2225 (2)	3.0 (2)
C(10)	6540 (2)	1269 (2)	3042 (2)	3.2 (2)
C(11)	4592 (5)	426 (2)	2096 (2)	3.9 (2)
C(12)	4015 (6)	375 (2)	1274 (2)	4.2 (2)
C(13)	3068 (5)	1206 (2)	999 (2)	3.4 (2)
C(14)	4020 (5)	2075 (2)	1217 (2)	3.0 (2)
C(15)	2613 (5)	2811 (2)	1156 (2)	3.4 (2)
C(16)	867 (5)	2321 (3)	1179 (2)	4.1 (2)
C(17)	1299 (5)	1348 (3)	1352 (2)	4.2 (2)
C(20)	5088 (5)	1287 (3)	3637 (2)	4.6 (2)
C(23)	2880 (6)	1141 (3)	128 (2)	4.5 (2)
C(24)	8048 (6)	2467 (3)	4500 (2)	5.1 (2)
C(25)	2784 (5)	3380 (2)	478 (2)	4.1 (2)
C(34)	10258 (5)	2895 (3)	3536 (2)	3.7 (2)
N(35)	2912 (5)	3814 (2)	-41 (2)	6.1 (2)
O(44)	11099 (4)	2884 (2)	2969 (2)	6.8 (2)
O(54)	10491 (4)	3534 (2)	4049 (1)	4.9 (2)
C(64)	11820 (7)	4184 (3)	3882 (3)	6.4 (3)

Table 2. Bond distances (Å), bond angles (°) and main torsion angles (°)

Averaged e.s.d.'s are respectively 0.005 Å, 0.3 and 0.7°.

C(1)—C(2)	1.532	C(9)—C(11)	1.524
C(1)—C(10)	1.527	C(10)—C(20)	1.549
C(2)—C(3)	1.515	C(11)—C(12)	1.526
C(3)—C(4)	1.549	C(12)—C(13)	1.527
C(4)—C(5)	1.567	C(13)—C(14)	1.549
C(4)—C(24)	1.541	C(13)—C(17)	1.534
C(4)—C(34)	1.513	C(13)—C(23)	1.553
C(5)—C(6)	1.533	C(14)—C(15)	1.561
C(5)—C(10)	1.556	C(15)—C(16)	1.552
C(6)—C(7)	1.529	C(16)—C(17)	1.528
C(7)—C(8)	1.533	C(25)—N(35)	1.130
C(8)—C(9)	1.540	C(34)—O(44)	1.201
C(8)—C(14)	1.545	C(34)—O(54)	1.332
C(9)—C(10)	1.576	O(54)—C(64)	1.455
C(2)—C(1)—C(10)	113.4	C(5)—C(10)—C(20)	114.0
C(1)—C(2)—C(3)	111.7	C(9)—C(10)—C(20)	109.6
C(2)—C(3)—C(4)	112.6	C(9)—C(11)—C(12)	110.9
C(3)—C(4)—C(5)	108.2	C(11)—C(12)—C(13)	114.0
C(3)—C(4)—C(24)	111.3	C(12)—C(13)—C(14)	111.9
C(3)—C(4)—C(34)	105.1	C(12)—C(13)—C(17)	114.9
C(5)—C(4)—C(24)	114.8	C(12)—C(13)—C(17)	114.9
C(5)—C(4)—C(34)	106.1	C(12)—C(13)—C(23)	108.2
C(24)—C(4)—C(34)	110.7	C(14)—C(13)—C(17)	102.5
C(4)—C(5)—C(6)	112.4	C(14)—C(13)—C(23)	110.2
C(4)—C(5)—C(10)	116.7	C(17)—C(13)—C(23)	109.1
C(6)—C(5)—C(10)	112.5	C(8)—C(14)—C(13)	115.1
C(5)—C(6)—C(7)	109.8	C(8)—C(14)—C(15)	108.7
C(6)—C(7)—C(8)	112.2	C(13)—C(14)—C(15)	103.7
C(7)—C(8)—C(9)	110.9	C(14)—C(15)—C(16)	106.6
C(7)—C(8)—C(14)	108.7	C(14)—C(15)—C(25)	113.5
C(9)—C(8)—C(14)	112.9	C(16)—C(15)—C(25)	111.9
C(8)—C(9)—C(10)	112.2	C(15)—C(16)—C(17)	105.2
C(8)—C(9)—C(11)	109.4	C(13)—C(17)—C(16)	104.5
C(10)—C(9)—C(11)	113.6	C(15)—C(25)—N(35)	179.9
C(1)—C(10)—C(5)	107.9	C(4)—C(34)—O(44)	125.4
C(1)—C(10)—C(9)	108.9	C(4)—C(34)—O(54)	114.2
C(1)—C(10)—C(20)	110.1	O(44)—C(34)—O(54)	120.4
C(5)—C(10)—C(9)	106.3	C(34)—O(54)—C(64)	116.1
C(10)—C(1)—C(2)—C(3)	-57.8	C(14)—C(8)—C(9)—C(11)	-54.7
C(1)—C(2)—C(3)—C(4)	56.9	C(8)—C(9)—C(11)—C(12)	59.7
C(2)—C(3)—C(4)—C(5)	-52.3	C(9)—C(11)—C(12)—C(13)	-57.7
C(3)—C(4)—C(5)—C(10)	51.7	C(11)—C(12)—C(13)—C(14)	47.5
C(4)—C(5)—C(10)—C(1)	-52.0	C(12)—C(13)—C(14)—C(8)	-42.1
C(5)—C(10)—C(1)—C(2)	53.0	C(13)—C(14)—C(8)—C(9)	46.9
C(10)—C(5)—C(6)—C(7)	-59.6	C(17)—C(13)—C(14)—C(15)	-37.0
C(5)—C(6)—C(7)—C(8)	55.5	C(13)—C(14)—C(15)—C(16)	18.6
C(6)—C(7)—C(8)—C(9)	-54.4	C(14)—C(15)—C(16)—C(17)	7.0
C(7)—C(8)—C(9)—C(10)	56.1	C(15)—C(16)—C(17)—C(13)	-30.4
C(8)—C(9)—C(10)—C(5)	-57.3	C(16)—C(17)—C(13)—C(14)	42.2
C(9)—C(10)—C(5)—C(6)	59.3		

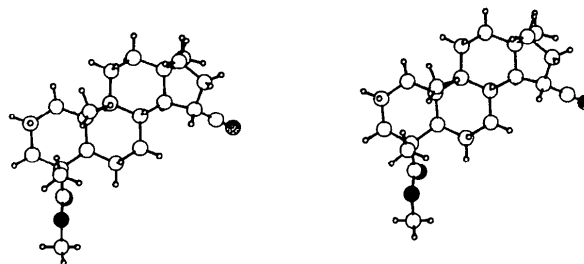


Fig. 2. Stereodrawing of the molecule.

References

BUCOURT, R. (1964). *Bull. Soc. Chim. Fr.* pp. 2080–2092.  
 CURINI, M., COCCIA, R., CECCHERELLI, P., HALLS, T. D. J.,  
 PORTER, B. & WENKERT, E. (1984). *J. Org. Chem.* **49**,  
 2054–2056.

- GILMORE, C. J. (1984). *J. Appl. Cryst.* **17**, 42–46.  
*International Tables for X-ray Crystallography* (1974). Vol. IV.  
 Birmingham: Kynoch Press. (Present distributor D. Reidel,  
 Dordrecht.)
- PRECIGOUX, G., BUSETTA, B., COURSEILLE, C. & HOSPITAL, M.  
 (1972). *Cryst. Struct. Commun.* **1**, 265–268.
- PRECIGOUX, G. & FORTNIER-MARQUINA, J. (1973). *Cryst. Struct.  
 Commun.* **2**, 287–290.
- STEINDL, H. & HASLINGER, E. (1985). *J. Org. Chem.* **50**,  
 3749–3752.
- STEWART, R. F., DAVIDSON, E. R. & SIMPSON, W. T. (1965). *J.  
 Chem. Phys.* **42**, 3175–3187.
- TAHARA, A., SHIMAYAKI, M., ITOH, M., YOSHIROTO, H. &  
 MASAYUKI, O. (1975). *Chem. Pharm. Bull.* **23**, 3189–3202.
- WEEKS, C. M., COOPER, A. & NORTON, D. A. (1971). *Acta Cryst.*  
**B27**, 1562–1572.
- WENKERT, E., DAVIS, L. L., MYHARI, B. L., SOLOMON, M. F., DA  
 SILVA, R. R., SHULMAN, S., WAMET, R. J., CECCHERELLI, P.,  
 CURINI, M. & PELLICCIARI, R. (1982). *J. Org. Chem.* **47**,  
 3242–3247.
- WHEELER, D. M. S. & WITT, P. R. (1972). *J. Org. Chem.* **37**,  
 4211–4214.
- WIRTHLIN, T., WEHRLI, H. & JEGGER, O. (1974). *Helv. Chim.  
 Acta*, **57**, 368–370.

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## Structure of 4-Phthalimido-*N*-(1,3-thiazol-2-yl)benzenesulfonamide

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**Abstract.** C<sub>17</sub>H<sub>11</sub>N<sub>3</sub>O<sub>4</sub>S<sub>2</sub>,  $M_r = 385.4$ , monoclinic,  $P2_1/c$ ,  $a = 7.955$  (2),  $b = 6.501$  (1),  $c = 35.859$  (8) Å,  $\beta = 116.92$  (2)°,  $V = 1653.5$  (5) Å<sup>3</sup>,  $Z = 4$ ,  $D_m = 1.560$ ,  $D_x = 1.548$  Mg m<sup>-3</sup>,  $\lambda(\text{Cu K}\alpha) = 1.5418$  Å,  $\mu = 3.13$  mm<sup>-1</sup>,  $F(000) = 792$ ,  $T = 293$  K, final  $R = 0.061$  for 2116 observed reflections. The molecule exists as an *imido* tautomer with deprotonation of the sulfonamide nitrogen and protonation of the thiazole nitrogen. Centrosymmetrically related pairs of molecules dimerize through hydrogen bonds of the type N–H···O and N–H···N. The thiazole and the phthalimide rings form stacks amongst themselves.

**Introduction.** Sulfonamides and their derivatives are well known antibacterial drugs. The present compound is a substituted sulfonamide with the substituents a thiazole at the sulfonamide nitrogen, N(1), and a phthalimide ring at the *para* position of the phenyl ring, and is used in the treatment of intestinal infections. The main advantage of this drug is that it does not build up in the blood because of its rapid excretion and poor absorption and hence large doses can be administered without danger of toxic effects. The X-ray structural study of the present compound has been carried out in continuation of our studies (Basak, Mazumdar & Chaudhuri, 1982, 1983, 1984; Basak, Chaudhuri & Mazumdar, 1984) of the substituted sulfonamides in order to obtain detailed structural information and also to study the effects of substituents on the molecular

geometry and the change in conformations with the ultimate aim of obtaining a better insight into the activity of these compounds.

**Experimental.** Transparent, slightly yellowish rectangular crystals from a mixture of acetone and water (1:1), density by flotation (benzene–bromoform), symmetry from oscillation and Weissenberg photographs,  $P2_1/c$  (absences:  $0k0$ ,  $k$  odd;  $h0l$ ,  $l$  odd), crystal size:  $0.23 \times 0.18 \times 0.15$  mm, 5051 equivalent reflections collected, Enraf–Nonius CAD-4 diffractometer, moving crystal/moving counter technique,  $2 \leq 2\theta \leq 150^\circ$ ,  $-8$  to  $8$  ( $h$ ),  $0$  to  $7$  ( $k$ ),  $-40$  to  $40$  ( $l$ ). Cell parameters from 32 high-angle reflections ( $23 \leq \theta \leq 39^\circ$ ),  $\bar{3}, 0, 12$  used as intensity standard, average count (= 1361) with  $\sigma(\text{calc.}) = 106$  (7.8%), 2130 reflections were non-zero after merging ( $R = 0.0909$ ) with  $I \geq 2.5\sigma(I)$ . No absorption corrections were applied. Structure solved by direct methods using *SHELX76* (Sheldrick, 1976) with  $E$ 's  $\geq 1.475$ . Anisotropic full-matrix least-squares refinement based on  $F$  with H's (from  $\Delta F$  synthesis) isotropic gave  $R = 0.068$ .

Further refinement, after omitting 14 reflections with  $(F_o - F_c)/\sigma \geq 3.0$  (poor agreement probably due to poor quality of the crystals or extinction), gave final  $R = 0.061$ ,  $wR = 0.063$ ,  $R_G = 0.079$ ,  $S = 2.028$ , with  $w = 1/\sigma^2(|F_o|) + 0.008512(F)^2$ . In the final cycle, maximum shift/ $\sigma$  is 0.1, average shift/ $\sigma < 0.04$ ; peak heights in final  $\Delta F$  map from  $-0.28$  to  $0.32$  e Å<sup>-3</sup>.